

SHORT  
COMMUNICATIONSReaction of 4-Aryl-4-oxobutane-1,1,2,2-tetracarbonitriles  
with Hydrochloric AcidO. V. Ershov, K. V. Lipin, V. N. Maksimova, A. V. Eremkin,  
Ya. S. Kayukov, and O. E. NasakinI.N. Ul'yanov Chuvash State University, Moskovskii pr. 15, Cheboksary, 428015 Russia  
e-mail: oleg.ershov@mail.ru

Received June 10, 2008

DOI: 10.1134/S1070428009030269

We previously reported that tetracyanoalkanones react with hydrohalic acids to give alkyl-substituted 2-halopyridine-3,4-dicarbonitriles [1] or 2-halo-6-hydroxy-1,4,5,6-tetrahydropyridine-3,4,4-tricarbonitriles [2]. In the present communication we describe a novel path of this reaction, which is possible only with aryl-substituted tetracyanoalkanones. 4-Aryl-4-oxobutane-1,1,2,2-tetracarbonitriles **Ia–Ic** were found to react with hydrochloric acid to produce either 6-aryl-2-chloropyridine-3,4-dicarbonitriles **IIa–IIc** or 3a-(2-aryl-2-oxoethyl)perhydropyrrolo[3,4-*c*]pyrrole-1,3,4,6-tetraones **IIIa–IIIc**.

No data were reported previously on the effect of water on reactions of tetracyanoalkanones with hydrogen halides. According to our results, this factor is very important in the reactions with aryl-substituted tetracyanoalkanones: pyridines **IIa–IIc** are formed in anhydrous medium, whereas diimides **IIIa–IIIc** are obtained in the presence of water. The structure of compounds **IIa–IIc** and **IIIa–IIIc** was confirmed by the IR, <sup>1</sup>H NMR, and mass spectra.

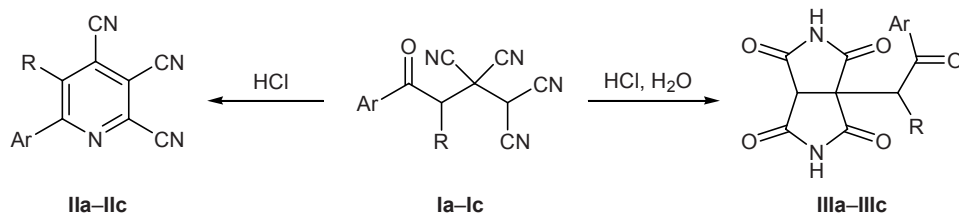
**2-Chloro-6-(4-methoxyphenyl)pyridine-3,4-dicarbonitrile (IIa).** Acetyl chloride, 1.5–2 ml, was added to a suspension of 0.56 g (2 mmol) of 4-(4-methoxyphenyl)-4-oxobutane-1,1,2,2-tetracarbo-

nitrile (**Ia**) in propan-2-ol. After some time, the precipitate was filtered off and washed with water and propan-2-ol. Yield 0.41 g (76%), mp 133–134°C. IR spectrum:  $\nu$  2228 cm<sup>-1</sup> (C≡N). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 3.87 s (3H, OCH<sub>3</sub>), 7.14 d and 8.21 d (2H each, C<sub>6</sub>H<sub>4</sub>, *J* = 9 Hz), 8.79 s (1H, 5-H). Mass spectrum: *m/z* 269 (*I*<sub>rel</sub> 100%). Found, %: C 62.75; H 2.79; N 14.89. C<sub>14</sub>H<sub>8</sub>ClN<sub>3</sub>O. Calculated, %: C 62.35; H 2.99; N 15.58.

Compounds **IIb** and **IIc** were synthesized in a similar way.

**2-Chloro-6-(3,4-dimethoxyphenyl)pyridine-3,4-dicarbonitrile (IIb).** Yield 0.47 g (78%), mp 139–140°C. IR spectrum:  $\nu$  2221 cm<sup>-1</sup> (C≡N). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 3.87 s and 3.88 s (3H each, OCH<sub>3</sub>), 7.17 d (1H, H<sub>arom</sub>, *J* = 9 Hz), 7.74 s (1H, H<sub>arom</sub>, *J* = 9 Hz), 7.89 d (1H, H<sub>arom</sub>, *J* = 9 Hz), 8.86 s (1H, 5-H). Mass spectrum: *m/z* 299 (*I*<sub>rel</sub> 100%). Found, %: C 59.75; H 3.45; N 13.42. C<sub>15</sub>H<sub>10</sub>ClN<sub>3</sub>O<sub>2</sub>. Calculated, %: C 60.11; H 3.36; N 14.02.

**2-Chloro-6-(4-methoxyphenyl)-5-methylpyridine-3,4-dicarbonitrile (IIc).** Yield 0.40 g (75%), mp 141–142°C. IR spectrum:  $\nu$  2228 cm<sup>-1</sup> (C≡N). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 2.59 s (3H, CH<sub>3</sub>), 3.87 s (3H, OCH<sub>3</sub>), 7.11 d and 7.65 d (2H each, C<sub>6</sub>H<sub>4</sub>, *J* = 9 Hz). Mass spectrum: *m/z* 283 (*I*<sub>rel</sub> 100%). Found, %:



R = H, Ar = 4-MeOC<sub>6</sub>H<sub>4</sub> (**a**), 3,4-(MeO)<sub>2</sub>C<sub>6</sub>H<sub>3</sub> (**b**); R = Me, Ar = 4-MeOC<sub>6</sub>H<sub>4</sub> (**c**).

C 63.78; H 3.55; N 14.81.  $C_{15}H_{10}ClN_3O$ . Calculated, %: 63.50; H 3.75; N 15.01.

**3a-[2-(4-Methoxyphenyl)-2-oxoethyl]perhydropyrrolo[3,4-c]pyrrole-1,3,4,6-tetraone (IIIa).** Concentrated hydrochloric acid, 10 ml, was added under stirring to a suspension of 0.56 g (2 mmol) of compound **Ia** in propan-2-ol. After 2–3 h, the precipitate was filtered off and washed with water and propan-2-ol. Yield 0.52 g (81%), mp 286–288°C (decomp.). IR spectrum,  $\nu$ ,  $cm^{-1}$ : 3309, 3184 (NH); 1670–1750 (C=O).  $^1H$  NMR spectrum,  $\delta$ , ppm: 3.85 s (3H,  $OCH_3$ ), 3.91 s (2H,  $CH_2$ ), 3.93 s (1H, CH), 7.05 d and 7.94 d (2H each,  $C_6H_4$ ,  $J = 9$  Hz), 11.82 s (2H, NH). Mass spectrum:  $m/z$  316 ( $I_{rel}$  2%). Found, %: C 56.06; H 3.49; N 8.98.  $C_{15}H_{12}N_2O_6$ . Calculated, %: C 56.96; H 3.82; N 8.86.

Compounds **IIIb** and **IIIc** were synthesized in a similar way.

**3a-[2-(3,4-Dimethoxyphenyl)-2-oxoethyl]perhydropyrrolo[3,4-c]pyrrole-1,3,4,6-tetraone (IIIb).** Yield 0.53 g (76%), mp 231–232°C (decomp.). IR spectrum,  $\nu$ ,  $cm^{-1}$ : 3308, 3175 (NH); 1669–1747 (C=O).  $^1H$  NMR spectrum,  $\delta$ , ppm: 3.81 s and 3.85 s (3H each,  $OCH_3$ ), 3.93 s (2H,  $CH_2$ ), 3.94 s (1H, CH), 7.07 d (1H,  $H_{arom}$ ,  $J = 8$  Hz), 7.42 s (1H,  $H_{arom}$ ), 7.65 d (1H,  $H_{arom}$ ,  $J = 8$  Hz), 11.81 s (2H, NH). Mass spectrum:  $m/z$  346 ( $I_{rel}$  4%). Found, %: C 56.15; H 4.12; N 8.34.  $C_{16}H_{14}N_2O_7$ . Calculated, %: C 55.49; H 4.07; N 8.09.

**3a-[2-(4-Methoxyphenyl)-1-methyl-2-oxoethyl]perhydropyrrolo[3,4-c]pyrrole-1,3,4,6-tetraone (IIIc).** Yield 0.47 g (75%), mp 275–277°C (decomp.).

IR spectrum,  $\nu$ ,  $cm^{-1}$ : 1667–1751 (C=O), 3225 (NH).  $^1H$  NMR spectrum,  $\delta$ , ppm: 1.21 d (3H,  $CH_3$ ,  $J = 7$  Hz), 3.86 s (3H,  $OCH_3$ ), 4.37 s (1H, 6a-H), 4.45 q (1H,  $CHCH_3$ ,  $J = 7$  Hz), 7.07 d and 7.97 d (2H each,  $C_6H_4$ ,  $J = 9$  Hz), 11.73 s and 12.00 s (1H each, NH). Mass spectrum:  $m/z$  330 ( $I_{rel}$  2%). Found, %: C 61.29; H 4.67; N 9.34.  $C_{16}H_{14}N_2O_6$ . Calculated, %: C 58.18; H 4.27; N 8.48.

The IR spectra were recorded from samples dispersed in mineral oil on an FSM-1202 spectrometer with Fourier transform. The  $^1H$  NMR spectra were measured on a Bruker DRX-500 instrument at 500.13 MHz using  $DMSO-d_6$  as solvent and tetramethylsilane as internal reference. The mass spectra (electron impact, 70 eV) were obtained on a Finnigan MAT Incos-50 spectrometer. The purity of the products was checked by TLC on Silufol UV-254 plates; spots were visualized under UV light, by treatment with iodine vapor, or by thermal treatment.

This study was performed under financial support by the Russian Foundation for Basic Research (project no. 08-03-97015r-Povolzh'e-a). The authors thank the Nanotechnology Collective Use Center of the Chuvash Republic for recording the IR and mass spectra.

## REFERENCES

1. Nasakin, O.E., Nikolaev, E.G., Terent'ev, P.B., Bulai, A.Kh., and Lavrent'eva, I.V., *Khim. Geterotsykl. Soedin.*, 1987, p. 653.
2. Sheverdov, V.P., Ershov, O.V., Nasakin, O.E., Chernushkin, A.N., and Tafeenko, V.A., *Russ. J. Org. Chem.*, 2002, vol. 38, p. 1001.